

**M. TECH.**  
**FIRST SEMESTER EXAMINATION, 2015-2016**  
**INSTRUMENTAL METHOD OF ANALYSIS**

**Time:-3 Hours]****[Max Marks: 100**

*Note: Attempt all five questions.*

**Q. 1. Attempt any two of the following:**

[10x2]

- (a) Write an explanatory note on the scope and applications of analytical chemistry. Outline the steps involved in the analytical process.
- (b) List different type equilibria in aqueous media. Explain the concept of pH and the function of glass electrodes.
- (c) Explain the principle of UV-visible spectrophotometer. How the standard curves are prepared using it. Explain whether it can be used for determination of turbidity.

**Q. 2. Attempt any two of the following:**

[10x2]

- (a) State Beer-Lambert Law. What are its limitations and deviations?
- (b) What do you understand by X-Ray fluorescence spectroscopy? With a schematic diagram, list the major components of the X-Ray fluorescence spectrometer.
- (c) Write a short note on the different excitation methods employed in emission spectroscopy.

**Q. 3. Attempt any two of the following:**

[10x2]

- (a) Give a detailed account on various chromatographic parameters and their significance.
- (b) Draw a schematic diagram of High performance liquid chromatography (HPLC) and describe the components (including columns) and their functions.
- (c) Discuss the principle involved in atomic absorption spectrometry. What is the necessity of background correction in AAS? How is it carried out?

**Q. 4. Attempt any two of the following:**

[10x2]

- (a) Describe the functional aspects of the SEM with the help of a block diagram of the instrument.
- (b) Write down the principle of EPR and discuss its applications. Discuss the limitation of EPR.
- (c) Discuss the theory of Nuclear Magnetic Resonance (NMR). What are the major components of CW NMR spectroscopy?

**Q. 5. Attempt any four of the following:**

[5x4]

- (a) Atomization of a sample.
- (b) Different regions of electromagnetic spectrum.
- (c) Short note on Polarography.
- (d) TOC Analyzer.
- (e) Principle of Voltametry and Chronopotentiometry.
- (f) Principle of Gas chromatography.